

## Evan Chemistry Answers

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**balancing chemical equations - ap chemistry** - balancing chemical equations  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$  answer key balance the equations below: 1)  $\text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$  2)  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  3)  $2 \text{NaCl} + 1 \text{F}_2 \rightarrow 2 \text{NaF} + 1 \text{Cl}_2$  4)  $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  5)  $1 \text{Pb}(\text{OH})_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$  6)  $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2(\text{SO}_4)_3$  7)  $1 \text{CH}_4 + 2 \text{O}_2 \rightarrow 1 \text{CO}_2 + 2 \text{H}_2\text{O}$  8)  $1 \text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$  9)  $2 \text{C}_8\text{H}_{18} + 25 \text{O}_2 \rightarrow 16 \text{CO}_2 + 18 \text{H}_2\text{O}$  10)  $1 \text{FeCl}_3 + 3 \text{NaOH} \rightarrow 1 \dots$

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reaction be spontaneous at room temperature?

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